

Al-Nd-Ti (Aluminum-Neodymium-Titanium)

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Recently, [2004Zho] determined an isothermal section at 500 °C for this ternary system, which depicts two ternary compounds found earlier by [1995Nie1,2].

Binary Systems

The Al-Nd phase diagram reassessed with new additional experimental input by [2005Gao] shows the following intermediate compounds: Nd₃Al (*D*0₁₉, Ni₃Sn-type hexagonal), Nd₂Al (*C*23, Co₂Si-type orthorhombic), NdAl (ErAl-type orthorhombic), NdAl₂ (*C*15, MgCu₂-type cubic), βNdAl₃ (stable between 1205 and 888 °C), αNdAl₃ (Ni₃Sn-type hexagonal), NdAl₄ or βNd₃Al₁₁ (*D*1₃, Al₄Ba-type tetragonal), and αNd₃Al₁₁ (αLa₃Al₁₁-type orthorhombic). Recently, [2006Sch] reviewed the Al-Ti phase diagram. The intermediate phases stable at 500 °C are: TiAl₃ (tetragonal, space group *I4/mmm*), TiAl₂ (HfGa₂-type tetragonal), TiAl (*L*1₀, AuCu-type tetragonal), and Ti₃Al (*D*0₁₉, Ni₃Sn-type hexagonal). There are no intermediate phases in the Nd-Ti system. At 500 °C, (αTi) and (αNd) show very little mutual solubility.

Ternary Compounds

Two ternary compounds are known in this system. NdTi₂Al₂₀ (denoted τ₁ here) is cubic, *Fd* $\bar{3}m$, *Z* = 8, *a* = 1.4704 nm [1995Nie1]. Nd₆Ti₄Al₄₃ (denoted τ₂ here) is hexagonal, *P*6₃/*mcm*, *Z* = 2, *a* = 1.1124 nm and *c* = 1.8069 nm [1995Nie2].

Ternary Isothermal Section

With starting metals of purity > 99.9 mass %, [2004Zho] arc-melted 150 alloys in Ar atm. The alloy samples were given a final anneal at 500 °C for 150 h and quenched in liquid nitrogen. The phase equilibria were studied with x-ray powder diffraction, scanning electron microscopy, and differential thermal analysis. The isothermal section at 500 °C constructed by [2004Zho] is shown in Fig. 1. The two ternary compounds τ₁ and τ₂ are present at 500 °C. The solubility of Ti in the binary phases of Nd₃Al, Nd₂Al and NdAl₂ is up to 2.4, 3.5, and 16.1 at.% respectively. The solubility of Nd in Ti, Ti₃Al, and TiAl is less than 1 at.%.

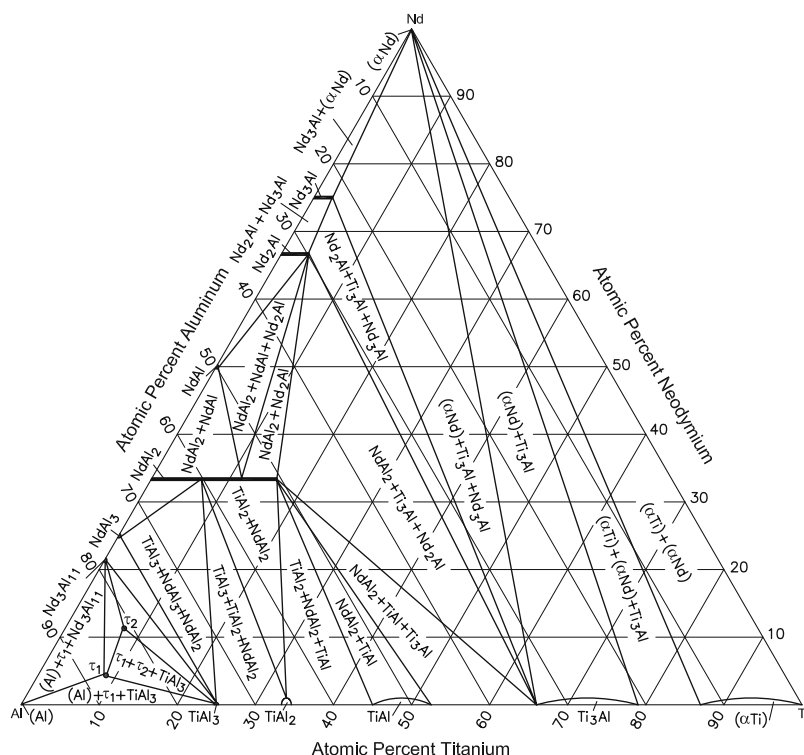


Fig. 1 Al-Nd-Ti isothermal section at 500 °C [2004Zho]. Narrow two-phase regions are omitted

References

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